Received July 7, 1994

present in solution. © 1994 Academic Press, Inc.

# EFFECT OF OZONATION PRODUCTS ON PHOSPHOLIPASE A<sub>2</sub> HYDROLYTIC ACTIVITY: USE OF BIS(1-HYDROXYHEPTYL)PEROXIDE AS A PRECURSOR OF THE OZONATION PRODUCT 1-HYDROXY-1-HYDROPEROXYHEPTANE

Maria Giulia Salgo, Giuseppe L. Squadrito, William A. Pryor\*

Biodynamics Institute, Louisiana State University, Baton Rouge, LA 70803-1800

Bis(1-hydroxyheptyl)peroxide (BisC7) upon spontaneous hydrolysis affords the
difficult-to-isolate ozonation product 1-hydroxy-1-hydroperoxyheptane along with
heptanal. PLA, hydrolytic activity is enhanced when the BisC7 hydrolysis products
are incorporated in the membrane of 1-palmitoyl-2-oleoyl-sn-glycero-3-
phosphocholine liposomes, suggesting they cause a pronounced alteration in the
bilayer packing order. Conversely, inhibition is observed when PLA <sub>2</sub> is incubated
with BisC7 hydrolytic products prior to incubation with the liposomes, suggesting

that these products are capable of reacting with and modifying the enzyme when

The activity of phospholipase A<sub>2</sub> (PLA<sub>2</sub>) is increased in tissues subjected to free radical damage as a consequence of oxidative stress. Preferential hydrolysis of peroxidized lipids has been observed in *in vivo* and *in vitro* systems (1-6) and it has also been shown that peroxidized lipids promote the hydrolysis of neighboring unoxidized lipids (7). Recently, we have shown that PLA<sub>2</sub> also can recognize ozone-induced damage (8).

Unsaturated fatty acids in phospholipids are important targets for reaction with ozone (9-11). The study of the effect of lipid ozonation products on PLA<sub>2</sub> hydrolytic activity is important since these molecules may be responsible for triggering an ozone-induced inflammatory response by activating the lipases PLA<sub>2</sub>

<u>Abbreviations</u>: BisC7, bis(1-hydroxyheptyl)peroxide;  $PLA_2$ , phospholipase  $A_2$ ; POPC, 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine.

<sup>\*</sup> To whom correspondence should be addressed at the Biodynamics Institute, 711 Choppin Hall, LSU, Baton Rouge, LA 70803-1800. FAX: (504) 388-4936.

and PLC. The ozonation of unsaturated fatty acids (UFA) yields primarily 1-hydroxy-1-hydroperoxyalkanes, aldehydes and Criegee ozonides (12-14). We have previously studied the effects of phospholipid Criegee ozonides and aldehydes on the hydrolytic activity of PLA<sub>2</sub> and found that they display remarkably different activities. While the Criegee ozonide of 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine is an excellent substrate for PLA<sub>2</sub>, its hydrolysis product, 1-palmitoyl-2-(9-oxononanoyl)-sn-glycero-3-phosphocholine, activates PLA<sub>2</sub> toward intact phospholipids within the same bilayer (8).

1-Hydroxy-1-hydroperoxyalkanes are difficult to isolate in pure form. However, they can be conveniently generated *in situ* from the spontaneous hydrolysis of the corresponding bis(1-hydroxyalkyl)peroxides which can be obtained as stable crystalline compounds. We recently succeeded in synthesizing bis(1-hydroxyheptyl)peroxide (BisC7). (The synthesis and characterization of BisC7, along with those of other ozonation products, will be published elsewhere; see equation 1 for the structure of BisC7.) Upon partial hydrolysis, BisC7 affords 1-hydroxy-1-hydroperoxyheptane (see equations 1-2), a major ozonation product from phospholipids containing palmitoleic acid. Here we report the change in PLA<sub>2</sub> hydrolytic activity caused by the hydrolysis products of BisC7, as reflected by oleic acid release from 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine (POPC) liposomes.

$$CH_{3}(CH_{2})_{5}CH HC(CH_{2})_{5}CH_{3} H_{2}O CH_{3}(CH_{2})_{5}CH + CH_{3}(CH_{2})_{5}CHO (1)$$

$$OH HO OH$$

$$Bis(1-hydroxyheptyl)peroxide (BisC7) CH_{2}OH$$

$$CH_{3}(CH_{2})_{5}CH H_{2}O CH_{3}(CH_{2})_{5}CHO + H_{2}O_{2} (2)$$

## **MATERIALS AND METHODS**

The following materials were used as received: L-a-1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine from Avanti Polar Lipids (Alabaster, AL); methyl sulfoxide from Aldrich (Milwaukee, WI); phospholipase A<sub>2</sub> (Crotalus Adamanteus; EC 3.1.1.4) from Sigma (St.Louis, MO); L-a-1-palmitoyl-2-oleoyl-[oleoyl-1-<sup>14</sup>C]-sn-glycero-3-phosphocholine sp.act. 58 mCi/mmol) from New England Nuclear Research Products (Boston, MA). The EcoLite liquid scintillation cocktail was purchased from ICN Biomedicals (Irvine, CA).

Liposome preparation. Liposomes were prepared by extrusion, using POPC as phospholipid core, following a procedure described previously (8). Briefly,

unlabeled POPC was mixed with the corresponding labeled [ $^{14}$ C]-POPC (0.0015  $\mu$ Ci  $^{14}$ C-POPC /mg POPC), vortexed and dried under a nitrogen stream and resuspended in 10 mM Tris/150 mM KCl buffer (pH 7.4), to a final concentration of 10 mg/mL. The phospholipid suspension was then extruded ten times through an extruder (Lipex Biomembranes, Vancouver, BC) using two 0.1 $\mu$ m pore polycarbonate membrane, under an argon pressure of 2000 KPa.

PLA<sub>2</sub> mediated hydrolysis. The time course of PLA<sub>2</sub> hydrolysis of intact POPC liposomes shows linear rates of hydrolysis over a 15 min period. After this period, the hydrolytic activity plateaus, with 5.5% of the total POPC having undergone hydrolysis, presumably because the unilamellar liposomes have been destroyed. Thus, we chose to allow PLA<sub>2</sub> to act upon the liposomes for a period of 20 min for all experiments.

Sample aliquots of 2 mg POPC/0.003  $\mu$ Ci [ $^{14}$ C]-POPC in 10 mM Tris/150 mM KCl buffer (pH 7.4) containing 15 mM CaCl $_2$ , were incubated in the presence of BisC7, which was added in 20  $\mu$ L DMSO to give a final concentration of 10  $\mu$ M, in a final volume of 1 mL, for 15, 30, 60, 90, 120 minutes at 37°C in a shaking bath. After each incubation period, 0.5 U of PLA $_2$  was added to each sample and incubated for an additional 20 min. Control experiments indicated that DMSO alone has no effect on PLA $_2$  hydrolytic activity.

In order to study the direct effect of the hydrolytic products of BisC7 on PLA<sub>2</sub>, sample aliquots containing 0.5 U PLA<sub>2</sub> were incubated with 10  $\mu$ M BisC7, for 15, 30, 60, 90, 120 min at 37°C in a shaking bath. After each incubation, 2 mg POPC/0.003  $\mu$ Ci [<sup>14</sup>C]-POPC, liposome aliquots, were added and incubated for an additional 20 min. A third group of samples, containing liposomes aliquots and BisC7 were incubated under identical conditions but in the absence of enzyme, to estimate the degree of nonenzymatic (spontaneous) hydrolysis.

The percentage of POPC hydrolysis catalyzed by PLA<sub>2</sub>, which was used as a standard activity, was measured following the release of [ $^{14}$ C]-oleic acid in a time-course assay, incubating liposomes (2 mg POPC/ 0.003  $\mu$ Ci [ $^{14}$ C]-POPC) for 0, 5, 10, 15, 20 min at 37  $^{\circ}$ C, under standard conditions.

The reactions were stopped by adding 5 mL of CHCl<sub>3</sub>/MeOH (2:1 v/v) followed by centrifugation and recovery of the organic phase. A second extraction was made using 3 mL of CHCl<sub>3</sub>. The organic phases were pooled and evaporated under a stream of nitrogen, the residue was dissolved in 3 mL of CHCl<sub>3</sub> and applied to a diol solid phase extraction column (Bond-Elute, Analychem) (25 mg, 3 mL). The chloroform fraction containing the free fatty acids, was collected into scintillation vials for measurements of radioactivity. The columns were then eluted with 3 mL of MeOH and the eluent collected in a separate set of vials, representing the phospholipid fraction. All samples were evaporated and redissolved in 10 mL of EcoLite liquid scintillation cocktail for measurement of radioactivity.

### **RESULTS AND DISCUSSION**

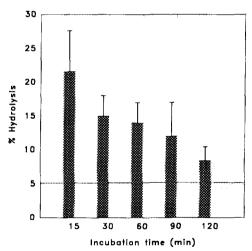
The inhibitory effects of the hydrolysis products of BisC7 on PLA<sub>2</sub> were evaluated by incubating PLA<sub>2</sub> with BisC7 for up to 120 min prior to the addition of liposomes (Table 1). The amount of oleic acid released decreases with the length of time in which BisC7 was incubated with PLA<sub>2</sub>, suggesting a direct inactivation of PLA<sub>2</sub> by hydrolysis products of BisC7.

Table 1. Percentage hydrolysis of oleic acid from POPC liposomes by phospholipase A<sub>2</sub> preincubated with bis(1-hydroxyheptyl)peroxide

Preincubation	ona (min) % Hyd	lrolysis <sup>c</sup>
Op	5.5 ±	0.25
15	4 ±	: 2
30	3 ±	2
60	1.8 ±	: 1
90	1.1 ±	0.4
120	0.46 ±	0.04

- a) Aliquots of liposomes containing 2 mg POPC/0.003  $\mu$ Ci [ $^{14}$ C]-POPC in 10 mM Tris/150 mM KCl buffer (pH 7.4) were preincubated for 15, 30, 60, 90, 120 min, in the presence of BisC7 (10  $\mu$ M), in a final volume of 1 mL at 37 °C in a shaking bath.
- b) Liposomes were not preincubated with BisC7.
- c) PLA<sub>2</sub> hydrolytic activity is expressed as percentage of [<sup>14</sup>C]-oleic acid released from POPC liposomes over a 20 min incubation period with PLA<sub>2</sub>. (See Materials and Methods for experimental details.)

Strikingly contrasting results were obtained as a result of a change in the order of addition. Liposomes were incubated with BisC7 (Figure 1) for up to 120 min, in the absence of  $PLA_2$ , to allow BisC7 to partially hydrolyze (according to



<u>Figure 1.</u> Phospholipase  $A_2$  hydrolysis of oleic acid from POPC liposomes preincubated with bis(1-hydroxyheptyl)peroxide. The horizontal line at 5.5% hydrolysis represents the enzymatic activity in the absence of bis(1-hydroxyheptyl)peroxide.

equations 1 and 2) and the products to incorporate into the liposome bilayers. Then, 0.5 U PLA<sub>2</sub> was added and the mixture incubated for an additional 20 min. Under these conditions, PLA<sub>2</sub>-mediated hydrolysis of oleic acid from POPC corresponded to 21 to 7.5% of POPC, a 1.4- to 3.8-fold increase in hydrolytic activity. This contrasts sharply with the decrease in hydrolytic activity that was observed when PLA<sub>2</sub> was incubated with BisC7 and then the liposomes added. Therefore, PLA<sub>2</sub> is activated when the hydrolysis products of BisC7 are incorporated into the bilayer. The hydrophobic hydrolysis products of BisC7 partition into the bilayer, generating heterogeneous microdomains that affect lipid organization and increase the area of the phospholipid-water interface, making the membrane more susceptible to PLA<sub>2</sub> attack (15). We previously observed a similar effect when another lipid ozonation product, 1-palmitoyl-2-(9-oxononanoyl)-sn-glycero-3-phosphocholine, was incorporated in POPC liposomes (8).

The decrease in hydrolytic activity, observed when liposomes are added last, occurs because the hydrolysis products of BisC7 react directly with PLA<sub>2</sub> in solution causing a decrease in oleic acid release (Table 1), suggesting the possibility of Schiff base formation at the lysine residue in the active site of PLA<sub>2</sub> (16) and/or noncovalent binding of 1-hydroxy-1-hydroperoxyheptane to the active site (17) and/or oxidations caused by 1-hydroxy-1-hydroperoxyheptane. When BisC7 is preincubated with the liposomes, the hydrophobic hydrolysis products preferentially partition into the bilayer, sharply lowering their effective concentrations in the aqueous phase where the PLA<sub>2</sub> resides, and thus, preventing direct reaction of the hydrolysis with the enzyme.

A decrease in PLA<sub>2</sub> hydrolytic activity was observed with longer times of BisC7 incubation with liposomes (Figure 1). This effect presumably occurs also as a result of the destabilizing effect that the incorporation of the hydrolysis products of BisC7 have on the liposomes bilayer. In the presence of the hydrolysis products of BisC7, and independently of PLA<sub>2</sub>, the unilamellar liposomes appear to have a faster intrinsic decay constant.

Thus, PLA<sub>2</sub> hydrolytic activity is influenced in two different ways by the hydrolysis products of bis(1-hydroxyheptyl)peroxide, one when the products are present in solution, and a second if they are embedded in the membrane. The hydrolysis products, namely 1-hydroxy-1-hydroperoxyheptane and heptanal, activate PLA<sub>2</sub> when embedded in the membrane, but inactivate PLA<sub>2</sub> when they are present in solution.

We propose long-chain 1-hydroxy-1-hydroperoxyalkanes will reside preferentially in the membrane and will activate PLA<sub>2</sub>, whereas the more water-soluble short-chain compounds will inactivate PLA<sub>2</sub>. The toxic effects of 1-hydroxy-1-hydroperoxyalkanes produced by the ozonation of unsaturated phospholipids during inhalation of ozone-containing air can be due to alteration of membrane packing order with consequent PLA<sub>2</sub> hydrolytic activation, and/or direct modification of the enzyme in solution, inhibiting its activity. Studies of the thermodynamics and kinetics of these processes will help understand the markedly different effects that have been observed for several lipid ozonation products on eicosanoid metabolism (18) and phospholipases A<sub>2</sub> and C (19) in airway epithelial cells.

#### **ACKNOWLEDGMENTS**

This work was supported by grants from the Health Effects Institute and from the National Institutes of Health.

#### REFERENCES

- 1. Borowitz, S.M., and Montgomery, C. (1989) Biochem. Biophys. Res. Commun. 158, 1021-1028.
- 2. Weglicki, W.B., Dickens, B.F., and Mak, I.T. (1984) Biochem. Biophys. Res. Commun. 124, 229-235.
- 3. Yasuda, M., and Fujita, T. (1977) Japan. J. Pharmacol. 27, 429-435.
- 4. Sevanian, A., Muakkassah-Kelly, S.F., and Montestruque, S. (1983) Arch. Biochem. Biophys.223, 441-452.
- 5. Salgo, M.G., Corongiu, F.P., and Sevanian, A. (1993) Arch. Biochem. Biophys. 304, 123-132.
- 6. Salgo, M.G., Corongiu, F.P., and Sevanian, A. (1992) Biochim. Biophys. Acta 1127, 131-140.
- 7. Sevanian, A., and Kim, E. (1985) J. Free Rad. Biol. Med. 1, 263-271.
- 8. Salgo, M.G., Squadrito, G.L., and Pryor, W.A. (1994) Chem. Res. Toxicol. 7, 458-462.
- Giamalva, D.H., Church, D.F., and Pryor, W.A. (1986) J. Am. Chem. Soc. 108, 6646-6651.
- Giamalva, D.H., Church, D.F., and Pryor, W.A. (1985) Biochem. Biophys. Res. Commun. 133, 773-779.
- 11. Uppu, R.M., and Pryor, W.A. (1992) Biochem. Biophys. Res. Commun. 187, 473-479.
- 12. Pryor, W.A., Das, B., and Church, D.F. (1991) Chem. Res. Toxicol. 4, 341-348.
- 13. Pryor, W.A., and Church, D.F. (1991) In Oxidative Damage & Repair: Chemical, Biological, and Medical Aspects (K.J.A. Davies, Ed.), pp. 496-504. Pergamon Press, New York, NY.

- 14. Squadrito, G.L., Uppu, R.M., Cueto, R., and Pryor, W.A. (1992) Lipids 27, 955-958.
- 15. Upreti, G.C., and Jain, M.K. (1980) J. Membr. Biol. 55, 113-121.
- 16. Wells, M.A. (1973) Biochemistry 12, 1086-1093.
- 17. Yu, L., and Dennis, E.A. (1992) J. Am. Chem. Soc. 114, 8757-8763.
- 18. Leikauf, G.D., Zhao, Q., Zhou, S., and Santrock, J. (1993) Am. J. Respir. Cell Mol. Biol. 9, 594-602.
- 19. Wright, D.T., Friedman, M., Pryor, W.A., Squadrito, G.L., Salgo, M.G., and Adler, K.B. (1994) Am. J. Respir. Crit. Care Med. 149, A320.